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### HIGHLIGHTS

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### ABSTRACT

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### 1. Introduction

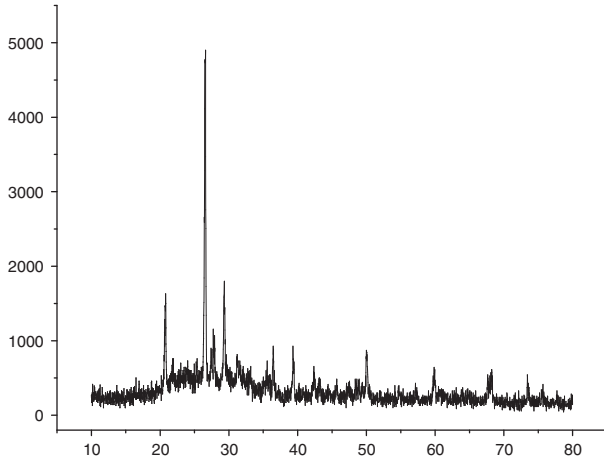
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Table 1

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### 3. Result and discussion

#### 3.1. Characterization of MSWI bottom ash

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#### 3.2. Leaching and retention behavior of nitrite

##### 3.2.1. Effect of the dosage

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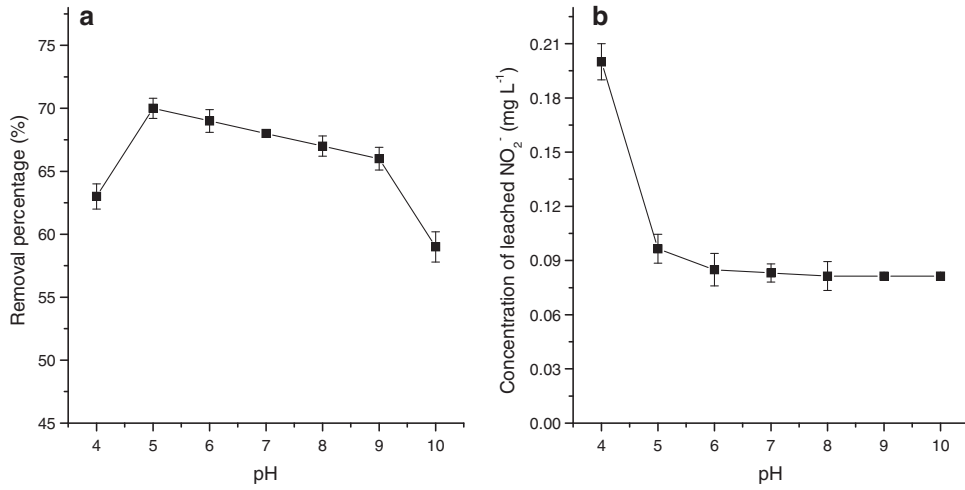


Fig. 3. (a) Removal percentage (%) vs pH. (b) Concentration of leached  $\text{NO}_2^-$  (mg L<sup>-1</sup>) vs pH.

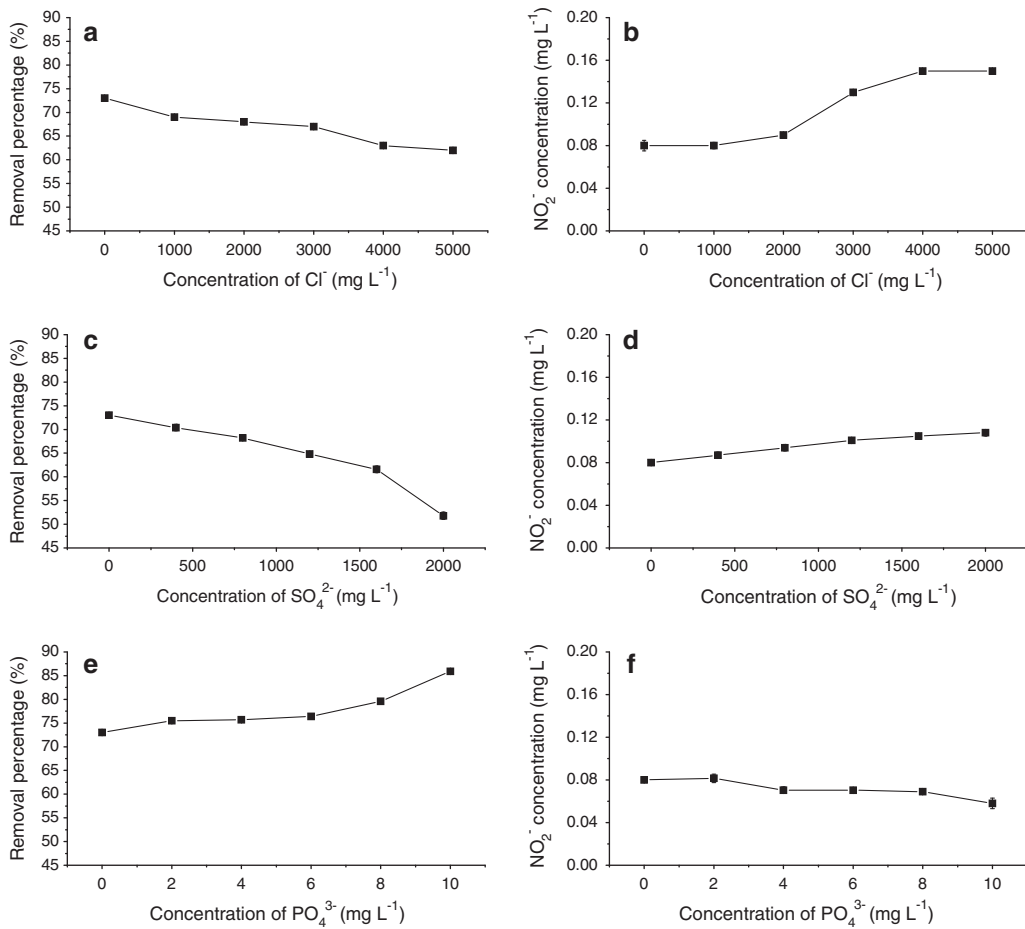
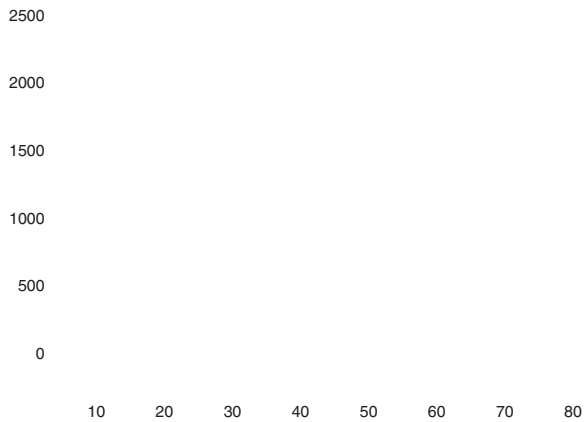


Fig. 4. (a) Removal percentage (%) vs  $\text{Cl}^-$  concentration (mg L<sup>-1</sup>). (b)  $\text{NO}_2^-$  concentration (mg L<sup>-1</sup>) vs  $\text{Cl}^-$  concentration (mg L<sup>-1</sup>). (c) Removal percentage (%) vs  $\text{SO}_4^{2-}$  concentration (mg L<sup>-1</sup>). (d)  $\text{NO}_2^-$  concentration (mg L<sup>-1</sup>) vs  $\text{SO}_4^{2-}$  concentration (mg L<sup>-1</sup>). (e) Removal percentage (%) vs  $\text{PO}_4^{3-}$  concentration (mg L<sup>-1</sup>). (f)  $\text{NO}_2^-$  concentration (mg L<sup>-1</sup>) vs  $\text{PO}_4^{3-}$  concentration (mg L<sup>-1</sup>).



### 3.2.2. Effect of pH

The effect of pH on the system is studied. The results show that as the pH increases, the concentration of the species increases linearly. This is observed in the graph above, where the y-axis represents the concentration and the x-axis represents the pH. The data points are approximately: (0, 0), (10, 250), (20, 500), (30, 750), (40, 1000), (50, 1250), (60, 1500), (70, 1750), (80, 2000).

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### 3.2.3. Effect of anions

The effect of anions on the system is studied. The results show that as the concentration of anions increases, the concentration of the species increases. This is observed in the graph above, where the y-axis represents the concentration and the x-axis represents the concentration of anions. The data points are approximately: (0, 0), (10, 250), (20, 500), (30, 750), (40, 1000), (50, 1250), (60, 1500), (70, 1750), (80, 2000).

## Appendix A. Supplementary material

## References

### 3.2.4. Effect of organic acid

## 4. Conclusions

## Acknowledgements